

HIGH ENERGY CATHODE MATERIALS FOR THE Li-ION BATTERIES IN THE APPLICATION OF PLUG-IN HYBRID ELECTRIC VEHICLES

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INTRODUCTION

Coprecipitated spherical $\text{Li}_{1.2}\text{Ni}_{0.2}\text{Mn}_{0.6}\text{O}_2$ materials are well-known for their large specific capacity ($>260 \text{ mAh/g}$), high charge/discharge voltage (2.0 V-4.6 V vs. Li) and high packing density [1]. When combined with $\text{Li}_4\text{Ti}_5\text{O}_{12}$ (1.5 V vs. Li) anode in a full cell, they can provide 2V battery systems that possess excellent cycling performance with high capacity and improved thermal properties. Therefore, high capacity cathode material such as $\text{Li}_{1.2}\text{Ni}_{0.2}\text{Mn}_{0.6}\text{O}_2$ is very desirable for rechargeable Li-ion batteries that can be used in applications such as plug-in hybrid electric vehicles (PHEVs) that require 20 to 40 mile electric drive with one battery charge.

EXPERIMENTAL

$\text{Li}_{1.2}\text{Ni}_{0.2}\text{Mn}_{0.6}\text{O}_2$ was principally made in two steps: 1) coprecipitation of the spherical $\text{Ni}_{0.25}\text{Mn}_{0.75}\text{CO}_3$ carbonate precursor. the precursor was coprecipitated from a nickel and manganese sulfate solution and a sodium carbonate solution in the presence of ammonia as a chelating agent in a continuous stirring tank reactor. The collected $\text{Ni}_{0.25}\text{Mn}_{0.75}\text{CO}_3$ powder was thoroughly washed, filtered, and dried in a vacuum oven at 120°C for 12 hours; 2) Lithiation of the coprecipitated $\text{Ni}_{0.25}\text{Mn}_{0.75}\text{CO}_3$ by mixing it with Li_2CO_3 and annealing the mix at 900°C for 24h.

RESULTS AND DISCUSSION

Fig.1 shows the discharge profile of $\text{Li}_{1.2}\text{Ni}_{0.2}\text{Mn}_{0.6}\text{O}_2$ when it is combined with $\text{Li}_4\text{Ti}_5\text{O}_{12}$ in a full cell at increasing current rates from 0.1C to 1C. The constructed full cell can be charged/discharged within 0.5 and 3.0V and shows around 240 mAh/g specific capacity at a constant discharging current density of 0.1C (charging and discharging for 10 hrs). When the current density increases, the specific capacity shows a very small decrease. Up to current density of 1C, the cell shows ~200 mAh/g capacity with 84% capacity retention.

Fig. 2 shows the cycling performance of $\text{Li}_{1.2}\text{Ni}_{0.2}\text{Mn}_{0.6}\text{O}_2/\text{Li}_4\text{Ti}_5\text{O}_{12}$ constructed cell at C/3 rate. The cell shows no capacity loss after 200 cycles and almost 100% coulombic efficiency in each cycle. After charging and discharging for 200 cycles at a constant current, about 225 mAh/g specific capacity was kept.

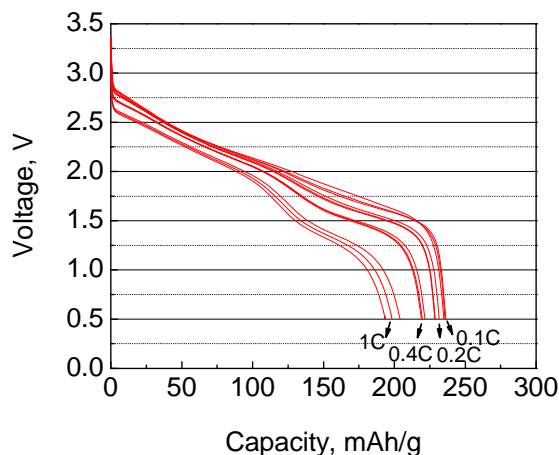


Fig.1. Discharge profile of $\text{Li}_{1.2}\text{Ni}_{0.2}\text{Mn}_{0.6}\text{O}_2/\text{Li}_4\text{Ti}_5\text{O}_{12}$ cell at increasing current rates (0.1C to 1C)

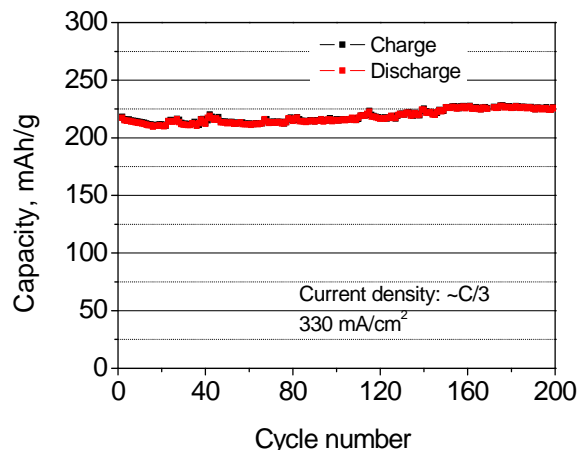


Fig.2. Cycling performance of $\text{Li}_{1.2}\text{Ni}_{0.2}\text{Mn}_{0.6}\text{O}_2/\text{Li}_4\text{Ti}_5\text{O}_{12}$ cell at C/3 rate.

REFERENCES

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